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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO
10/788,850	02/27/2004	Vincent J. Gatto	AN-7421	6851
7982 EDGAR SPIE	7590 10/24/2007 I.MAN		EXAMINER	
ALBEMARLE	CORPORATION		OH, TAYLOR V	
451 FLORIDA BLVD. BATON ROUGE, LA 70801		ART UNIT	PAPER NUMBER	
	- ·,		1625	
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Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

	Application No.	Applicant(s)			
	10/788,850	GATTO ET AL.			
Office Action Summary	Examiner	Art Unit			
	Taylor Victor Oh	1625			
The MAILING DATE of this communication app Period for Reply	ears on the cover sheet with the	e correspondence address			
A SHORTENED STATUTORY PERIOD FOR REPLY WHICHEVER IS LONGER, FROM THE MAILING DA  - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication.  - If NO period for reply is specified above, the maximum statutory period verallure to reply within the set or extended period for reply will, by statute Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	ATE OF THIS COMMUNICATION  36(a). In no event, however, may a reply be will apply and will expire SIX (6) MONTHS from the cause the application to become ABANDO	ON. timely filed om the mailing date of this communication. NED (35 U.S.C. § 133).			
Status					
1) Responsive to communication(s) filed on 15 A	<u>ugust 2007</u> .				
,	,				
closed in accordance with the practice under E	ex parte Quayle, 1935 C.D. 11,	453 O.G. 213.			
Disposition of Claims					
4) ⊠ Claim(s) <u>1-69</u> is/are pending in the application 4a) Of the above claim(s) <u>50-68</u> is/are withdray 5) ☐ Claim(s) is/are allowed. 6) ⊠ Claim(s) <u>1-49,69</u> is/are rejected. 7) ☐ Claim(s) is/are objected to. 8) ☐ Claim(s) are subject to restriction and/o	vn from consideration.				
Application Papers					
9) The specification is objected to by the Examine 10) The drawing(s) filed on is/are: a) acc Applicant may not request that any objection to the Replacement drawing sheet(s) including the correct 11) The oath or declaration is objected to by the Ex	epted or b) objected to by the drawing(s) be held in abeyance. Stion is required if the drawing(s) is	See 37 CFR 1.85(a). objected to. See 37 CFR 1.121(d).			
Priority under 35 U.S.C. § 119					
12) Acknowledgment is made of a claim for foreign a) All b) Some * c) None of:  1. Certified copies of the priority document 2. Certified copies of the priority document 3. Copies of the certified copies of the priority application from the International Bureau * See the attached detailed Office action for a list	is have been received. Is have been received in Applic rity documents have been rece u (PCT Rule 17.2(a)).	ation No ived in this National Stage			
Attachment(s)  1) Notice of References Cited (PTO-892)  2) Notice of Draftsperson's Patent Drawing Review (PTO-948)  3) Information Disclosure Statement(s) (PTO/SB/08)  Paper No(s)/Mail Date	4) Interview Summa Paper No(s)/Mail 5) Notice of Informa 6) Other:	Date			

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# Final Rejection

## **The Status of Claims**

Claims 1-69 are pending.

Claims 1-49 and 69 are rejected.

Claims 50-68 are withdrawn from consideration.

#### Claim Rejections - 35 USC § 112

The following is a quotation of the first paragraph of 35 U.S.C. 112:

The specification shall contain a written description of the invention, and of the manner and process of making and using it, in such full, clear, concise, and exact terms as to enable any person skilled in the art to which it pertains, or with which it is most nearly connected, to make and use the same and shall set forth the best mode contemplated by the inventor of carrying out his invention.

The rejection of Claims 1-4, and 7-37 under 35 U.S.C. 112, first paragraph has been withdrawn due to the modification made in the claims.

The rejection of Claims 38-49 under 35 U.S.C. 112, first paragraph, has been withdrawn due to the modification made in the claims.

The following is a quotation of the second paragraph of 35 U.S.C. 112:

The specification shall conclude with one or more claims particularly pointing out and distinctly claiming the subject matter which the applicant regards as his invention.

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The rejection of Claims 1-4, 7-37 and 43-49 under 35 U.S.C. 112, second paragraph has been withdrawn due to the modification made in the claims.

### Claim Rejections-35 USC 103

1. Applicants' argument filed 8/15/07 have been fully considered but they are not persuasive.

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

The rejection of Claims 1-49 and 69 are rejected under 35 U.S.C. 103(a) as being unpatentable over Haeberli et al (U.S. 4,228,297) in view of Grant et al (Chemical Dictionary, 1990, p. 11-12).

The rejection of Claims 1-49 and 69 are rejected under 35 U.S.C. 103(a) as being unpatentable over Haeberli et al (U.S. 4,228,297) in view of Grant et al (Chemical Dictionary, 1990, p. 11-12) has been maintained for the reasons of the record on 6/1/07.

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**Applicants' Argument** 

2. Applicants argue the following issues:

a. In the claims 28 and 31, the first and second catalysts are the same, which is not disclosed in the Haeberli et al;

- b. In claim 69, the claim requires a single catalyst to form a hindered pheolic alkylester; also, the phrase "consisting essentially of" in the claim does not allow the addition of ethyl alcohol to the process shown in the prior art;
- c. There is no motivation or suggestion in the Haeberli et al to use acids other than acetic acid in combination with Grant et al;
- d. there is no suggestion that the acid is added to neutralize the catalyst residues; furthermore, acidifying the reaction mixture with glacial acetic acid does not form a precipitated salt.

Applicants' arguments have been noted, but the arguments are not persuasive.

First, regarding the first and second arguments, the Examiner has noted applicants' argument. However, on the contrary to applicants' argument, the Haeberli et al expressly teaches that the first and second catalysts are the same as described below (see col. 9, lines 1-4):

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The second alkaline catalyst may be selected from the classes of compounds listed above for the first alkaline catalyst, i.e., alkali metal hydrides, alkali metal alkoxides of Formula VI or alkali metal amides of Formula

Furthermore, this means that the Haeberli et al requires the single catalyst to form a hindered phenolic alkylester as shown in the above.

Regarding the use of the phrase "consisting essentially of" in the claim languages, the MPEP 2111.03[R-3] defines in the following: the **consisting essentially of** claim occupies a middle ground between closed claims that are written in a consisting of format and fully open claims that are drafted in a comprising format. This implies that there is a room for other material to be present in the claim 69; this can be the organic solvent such as ethyl alcohol seen from the prior art. Therefore, the prior art is still relevant to the claimed invention.

Secondly, regarding the third and fourth arguments, the Examiner has noted applicants' argument. However, the acetic acid has the same function as the phosphoric acid in the claimed process with respect to the role of the acid in the neutralization step; regardless of the type of the acid to be used, the only role of the acid is to neutralize the base in the process. Furthermore, the phosphoric acid is one of the well-known acids in the art as shown in Grant et al (Chemical Dictionary, 1990, p. 11-12). Therefore, it would have been obvious to the skilled artisan in the art to be motivated to use the phosphoric acid as an alternative to acetic acid for the neutralization step in the prior art process. This is because the skilled artisan in the art

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would expect such a modification to be feasible and successful as shown in the prior art.

Moreover, regarding the failure of glacial acetic acid to form the precipitated salt, regardless of forming the precipitated salt by using glacial acetic acid, the Haeberli et al does disclose the removal of the catalyst in the following passages of the steps(see col. 10 ,lines 29-62):

three hours. The vacuum was then released with nitrogen, and the reaction mass was cooled to 70° C. and acidified with 3.0 g of glacial acetic acid. 132 g of ethyl alcohol were added to the melt, and the resultant solution was clarified. The filtrate was cooled to 28° C. and seeded with 0.5 g of thio-bis-{ethylene-3-(3,5-di-t-butyl-4-hydroxyphenyl)propionate}.

- The reaction product crystallized and the resulting slurry was cooled to 16° C. The product was isolated on a Buchner funnel, washed with cold ethyl alcohol, sucked dry and dried in a vacuum oven at 50° C. to a constant weight. 97.0 g of dry thio-bis-{ethylene-3-(3,5-
- di-t-butyl-4-hydroxyphenyl)propionate} were obtained; m.p. 71.5° C.; yield 75.4%, based on the thiodiglycol employed.

From this passage, it shows that, after the reaction mixture being acidified with acetic acid, the filtrate is cooled to promote the precipitation of the reaction product; the product was isolated on the funnel by washing with cold ethyl alcohol. This means that the salt of the catalyst can be removed by the filter. Therefore, the prior art process is still relevant to the claimed invention.

#### Conclusion

Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP

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§ 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Taylor Victor Oh whose telephone number is 571-272-0689. The examiner can normally be reached on 8:30-5:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Janet Andres can be reached on 571-272-0867. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

TAYLOR VICTOR OH PRIMARY EXAMINER

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